Improving the Photo-oxidative Performance of Bi₂MoO₆ by Harnessing the Synergy between Spatial Charge Separation and Rational Co-catalyst Deposition

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It has been repoted that photogenerated electrons and holes can be driven to specific crystal facets within a semiconductor, which is believed to arise from differences in their surface electronic structures, indicating different facets can act as either photo-reduction or phto-oxidation sites.^[1,2,3] For instance, the intrinsic nature of charge separation between the $\{010\}$ and $\{110\}$ facets of BiVO₄ has been demonstrated via the selective photo-deposition of metals and metal oxides.^[4] With this in mind, this work examined the electron-hole separation capabilities within faceted plate-like bismuth molybdate (Bi₂MoO₆), which is an efficient photocatalyst for water oxidation. Herein, we demonstrate that the photo-excited electrons and holes can be spatially separated to the $\{010\}$ and $\{001\}/\{100\}$ facets, respectively, within Bi_2MoO_6 as evidenced by the facet-dependant photo-deposition of noble metals (e.g. Pt, Au and Ag) or/and metal oxides (e.g. PbO₂, MnO_x, CoO_x), as shown in Fig. 1. Theoretical calculations revealed that differences in energy levels between the conduction band (CB) and valence band (VB) of the $\{010\}$ and $\{001\}/\{100\}$ facets can contribute to electrons and holes being driven to different surfaces of the plate-like Bi₂MoO₆ (Fig. 2). Utilising this knowledge, the photo-oxidative capability was then improved by adding an efficient water oxidation co-catalyst, CoO_x, to the system. Interestingly, we found that the degree of enhancement is highly dependent on the co-catalyst location. A greater amount of oxygen was evolved when CoO_x was selectively deposited on the oxidation (hole-rich) facets of Bi₂MoO₆ compared to Bi₂MoO₆ where the CoO_x was randomly distributed on the surface. The elevated performance exhibited when the CoO_x is selectively deposited on the Bi_2MoO_6 {001}/{100} facets is ascribed to the benefit of hole trapping by the co-catalyst being accentuated over other potentially detrimental co-catalyst effects such as them acting as recombination centres and/or covering reactive sites. These results indicate that the synergy between spatial charge separation and co-catalyst location on the appropriate facets of plate-like Bi₂MoO₆ can promote its photocatalytic activity.

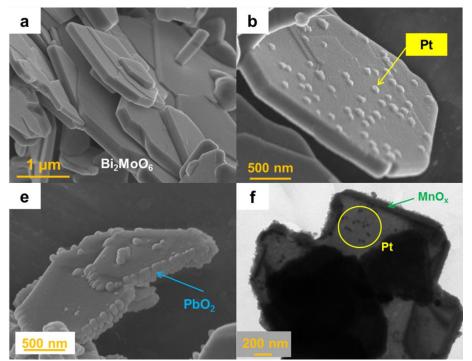


Figure 1. SEM images of (a) plate-like Bi_2MoO_6 , (b) photo-reduced Pt on Bi_2MoO_6 , (c) photo-oxidised PbO₂ on Bi_2MoO_6 , and (d) TEM image of Bi_2MoO_6 with dual photo-deposited MnO_x and Pt co-catalysts

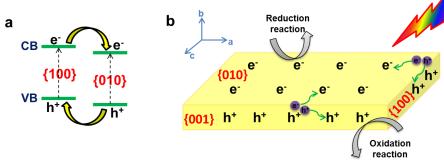


Figure 2. Schematic diagram of proposed charge separation process among $\{010\}$ and $\{001\}/\{100\}$ facets within plate-like Bi_2MoO_6

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