On the tautomerism of 2-mercaptobenzimidazol: A photoinduced reaction in solid argon

Elisa M. Brás and Rui Fausto

CQC, Department of Chemistry, University of Coimbra, P3004-535, Portugal

E-mail: elisabras7@gmail.com

Contrarily to what happens for the oxo-hydroxo tautomerism, the thione-thiol tautomerism has been only scarcely investigated in spite of its relevance in both chemical and biochemical [1,2].the present study, the thione-thiol phototautomerization In 2-mercaptobenzimidazole (2-MBI; Fig. 1) isolated in a cryogenic (15 K) argon matrix has been addressed using in situ narrowband UV irradiation of the matrix-isolated compound and infrared spectroscopy probing. The initially deposited matrix was found to contain exclusively the thione tautomer of **2-MBI**, which corresponds to its thermodynamically most stable form. Upon irradiation at $\lambda = 307$ nm, this tautomer converts into the thiol tautomer via intramolecular hydrogen atom transfer. Subsequent irradiation at $\lambda = 246$ nm selectively converts-back the thiol form into the thione tautomer, demonstrating the photoreversibility of the investigated photoprocess. The possibility of selectively convert each one of the tautomers of 2-MBI into the other one in an efficient way using optical excitation at different wavelengths makes this molecule a good candidate for investigation as a molecular switch.

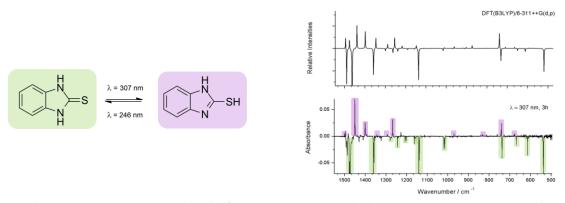


Figure 1. Observed phototautomerism in **2-MBI**. *Left:* observed photoprocesses and structures of the two tautomers. *Right:* infrared difference spectrum ($\lambda = 307$ nm irradiated *minus* as-deposited matrix) showing the thione—thiol transformation (*bottom*); calculated infrared difference spectrum (thiol *minus* thione) (*top*).

Funding: The CQC is supported by FCT, through the project UI0313/QUI/2013, co-funded by COMPETE UE. R. F. acknowledges the facilities and funds made available by the FCT project PTDC/QEQ-QFI/3284/2014 – POCI-01-0145-FEDER-016617.

Acknowledgement: We thank the ICP commission for the opportunity to present our work.

References:

- [1] S. Stoyanov, I. Petkov, L. Antonov, T. Stoyanova, P. Karagiannidis, P. Aslanidis, *Can. J. Chem.*, **1990**, 68, 1482
- [2] D. Moran, K. Sukcharoenphon, R. Puchta, H. Schaefer, P. Schleyer, C. Hoff, *J. Org. Chem.*, **2002**, 67 (25), 9061–9069