

## Dynamics simulations of photoinduced intramolecular proton transfer reactions of 2,5-bis(2'-benzoxazolyl)hydroquinone

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The dynamics simulations of the excited-state intramolecular proton transfer (ESIPT) of 2,5-bis(2'-benzoxazolyl)hydroquinone (BBHQ) have been performed on their lowest energy structures using time-dependent density functional theory (TDDFT) at TD-B3LYP/SVP level.<sup>[1]</sup> Our dynamics simulations showed that the intramolecular hydrogen bond N $\cdots$ H–O is strengthened in the excited state, which triggers driving force to effectively facilitate the proton transfer process through an intrinsic intramolecular hydrogen bond.<sup>[2,3,4]</sup> There are two possible mechanisms of the ESIPT processes: *a*) single proton transfer and *b*) double proton transfer. The ESIPT mechanism of BBHQ systems elucidates that single proton transfer is more likely to take place in comparison with the double proton transfer reaction.<sup>[4]</sup> Moreover, the detailed information of the mechanisms and photophysical properties of these systems will be discussed.

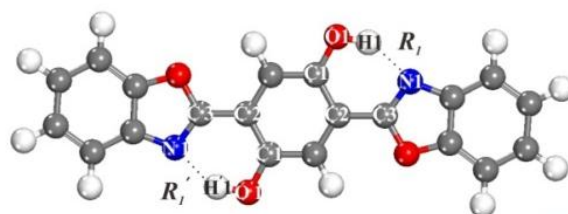


Figure 1. Optimized geometry of 2,5-bis(2'-benzoxazolyl)hydroquinone computed at B3LYP/SVP level.

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### References:

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